# **PCT**

#### WORLD INTELLECTUAL PROPERTY ORGANIZATION International Bureau



# INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification <sup>5</sup> :		(11) International Publication Number:	WO 94/04625
C09J 123/16	A1	(43) International Publication Date:	3 March 1994 (03.03.94
(21) International Application Number: PCT/US (22) International Filing Date: 20 August 1993 (		P.O. Box 5200, Baytown, TX	
(30) Priority data: 07/935,483 25 August 1992 (25.08.92)	) 1	(81) Designated States: CA, JP, Europe, DE, DK, ES, FR, GB, GR, SE).	
(71) Applicant: EXXON CHEMICAL PATENTS IN US]; 5200 Bayway Dr., Baytown, TX 77520 (U		Published With international search repo	rt.
(72) Inventors: ROBERTSON, Martha, Hetzel; Rour 169, Zachary, LA 70791 (US). BRANT, Patri Harborcrest, Seabrook, TX 77586 (US). McE Kenneth, Odel, Jr.; Clos des Ecossais 17, B-14 loo (BE).	ick ; 1 ELRAT	03   H,	

#### (57) Abstract

A hot melt adhesive including syndiotactic polypropylene having a polymer chain of at least 80 % racemic dyads and having a melting point of about 100  $^{\circ}$ C to 180  $^{\circ}$ C.

# FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	FR	France	MR	Mauritania
AU	Australia	GA	Gabon	MW	Malawi
BB	Barbados	GB	United Kingdom	NE	Niger
BE	Belgium	GN	Guinea	NL	Netherlands
BF	Burkina Faso	GR	Greece	NO	Norway
BG	Bulgaria	HU	Hungary	NZ	New Zealand
BJ	Benin	1E	Ireland	PL	Poland
BR	Brazil	ìТ	Italy	PT	Portugal
BY	Belarus	JР	Japan	RO	Romania
CA	Canada	KP	Democratic People's Republic	RU	Russian Federation
CF	Central African Republic		of Korea	SD	Sudan
CG	Congo	KR	Republic of Korea	SE	Sweden
CH	Switzerland	ΚZ	Kazakhstan	SI	Slovenia
CI	Côte d'Ivoire	LI	Liechtenstein	SK	Slovak Republic
CM	Cameroon	LK	Sri Lanka	SN	Senegal
CN	China	LU	Luxembourg	TD	Chad
CS	Czechosłovakia	LV	Latvia	TG	Togo
CZ	Czech Republic	MC	Monaco	UA	Ukraine
DE	Germany	MG	Madagascar	US	United States of America
DK	Denmark	ML	Mali	UZ	Uzbekistan
ES	Spain	MN	Mongolia	VN	Viet Nam
FI	Finland		-		

-1-

#### SYNDIOTACTIC HOT MELT ADHESIVE

5

# Field of the Invention

The invention relates to hot melt adhesives comprised of syndiotactic polymers and a tackifier.

10

15

20

25

30

35

# Background of the Invention

A class of adhesive compositions known as thermoplastic or hot melt adhesives have enjoyed continually increasing usage in many industrial applications. These adhesives, or hot melts, as they are commonly designated, are solid or semisolid combinations of film forming resins, tackifying resins, rubbery polymers, plasticizers, waxes and similar materials which are added to the adhesive composition in order to impart various properties thereto.

Adhesive bonds derived from hot melts are particularly useful because (1) they are 100 percent solid materials and (2) they quickly form tough adhesive bonds simply by cooling--i.e. no need for solvent evaporation, etc.

Adhesive derived from hot melts are particularly useful because of their very rapid set up or bonding time. Strong bonds are formed when the adhesive cools below its melt point and crystallizes. There is no long wait for solvent or water evaporation or for chemical curing as with other type adhesives. Depending on composition, hot melt adhesives can be prepared having good tack, high bond

strength, good flexibility, low temperature properties, good environmental resistance etc.

A typical class of hot melt adhesive composition utilizes polyolefin polymers as the base or carrier material. The polyolefin base is usually blended with other polymers and copolymers, resin tackifiers and with modifiers and additives.

# 10 Summary of the Invention

5

30

Therefore, it is one object of the invention to provide a hot melt adhesive having: (1) a high softening temperature; (2) improved toughness; (3) slow crystallization kinetics which result in

- longer "open times" between the application of the adhesives and the formation of the joint; or
  - (4) less shrinkage than corresponding isotactic polypropylene or ethylene/vinyl acetate based adhesives.
- These and other objects are accomplished by the present invention which provides a hot melt adhesive comprising syndiotactic polypropylene having a polymer chain including at least 80% racemic dyads and having a melting point of about 100°C to 180°C, preferably 100 to 160°C.

# Brief Description of the Drawings

Figure 1 is a graph depicting peel strength as a function of open time for an isotactic copolymer (17% hexene-1).

Figure 2 is a graph depicting peel strength as a function of open time for an syndiotactic copolymer (17% hexene-1).

Figure 3 and Figure 5 illustrate stressstrain curves on a C3/C6 copolymer/ECR-111 system containing 6% hexene.

-3-

Figure 4 depicts the results of a study where various controlled tacticity polypropylene (CTPP) films were heat sealed to each other with no adhesive sandwiched in the middle.

5

10

15

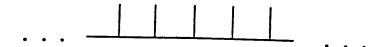
20

# Detailed Description of the Preferred Embodiments

The present invention comprehends making a hot-melt adhesive using syndiotactic polypropylene having a polymer chain including at least 80% racemic dyads, preferably 85%, and having a melting point between about 100°C and 180°C, preferably 100 to 160°C.

As is known in the art, the structure and properties of syndiotactic polypropylene (SPP) differ significantly from those of isotactic polypropylene. The isotactic structure is typically described as having the methyl groups attached to the tertiary carbon atoms of successive monomeric units on the same side of a hypothetical plane through the main chain of the polymer, e.g., the methyl groups are all above or below the plane. Using the Fischer projection formula the stereochemical sequence of isotactic polypropylene is described as follows:

25



30

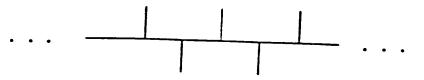
35

Another way of describing the structure is through the use of NMR. Bovey's NMR nomenclature for an isotactic pentad is ...mmmm... with each "m" representing a "meso" dyad of successive methyl groups on the same side of the plane. As known in the art, any deviation or inversion of the structure of the chain lowers the degree of isotacticity and crystallinity of the polymer.

In contrast to the isotactic structure, syndiotactic polymers are those in which the methyl groups attached to the tertiary carbon atoms of successive monomeric units in the chain lie on alternate sides of the plane of the polymer. Using the Fischer projection formula, the structure of a syndiotactic polymer is designated as:

10

5



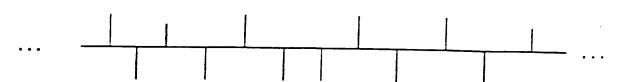
In NMR nomenclature, this pentad is described as
...rrr... in which each "r" represents a "racemic"
dyad, i.e. successive methyl groups on alternate sides
of the plane. The percentage of r dyads in the chain
determines the degree of syndiotacticity of the

polymer. This crystallinity distinguishes both
syndiotactic 1 and isotactic polymers from atactic
polymers. Atactic polymers exhibit no regular order of
repeating unit configurations in the polymer chain and
form essentially a non-crystalline product.

Most of the syndiotactic polypropylene produced previously as described in such references as U.S. Patents No. 3,305,538 and 3,258,455 to Natta et al have produced syndiotactic polymers having the following structure:

30

25



35

-5-

The polymer chain used in producing the hot melt adhesives of this invention preferably consists of greater than 80% racemic dyads, and more preferably greater than 85% racemic dyads. A method for producing SPP of this type is described in published

5

10

15

20

25

30

35

European Patent Application EP 351,391 A2, the entire disclosure of which is hereby incorporated by reference. Some catalysts which are suitable for producing the meso triad SPP of the present invention are described in U.S. Patent No. 4,892,851 to Ewen et al., the entire disclosure of which is hereby incorporated by reference.

An important feature of these syndiotactic polypropylenes is that they have lower heats of fusion than the corresponding isotactic polymers. Another important feature which distinguishes SPP from IPP is that SPP exhibits two endothermic peaks at temperatures greater than 100°C as determined by differential scanningcalorimetry.

The SPP polymers and copolymers of this invention may be blended with one or more tackifiers, such as aliphatic hydrocarbon resins prepared by the polymerization of monomers consisting primarily of olefins and diolefins and hydrogenated forms of these aliphatic hydrocarbon resins. These tackifying resins typically exhibit a Tg of -30 to 65 °C, and a ring and ball softening point of 12 to 100 °C. Examples include but not are not limited to hydrocarbon resins available under the tradenames ECR143H, and ESCOREZ 1310LC available from Exxon Chemical Company. tackifiers include the hydrocarbon products of the distillation of oil, particularly, hydrogenated cyclic resins with a Tg of -14\_to -70\_°C and a ring and ball softening point of 18 to 130 °C. Examples include but are not limited to those resins available from Exxon Chemical under the tradename ECR-327, ESCOREZ 5380 and

-6-

ESCOREZ 5320. These hydrogenated cyclic resins have the properties shown in Table X below.

TABLE X

5	Resin		Softening	Melt Viscosity	Mw	Mn	Mw/Mn	Tg C	
			Point°C	(cps)				(DSC)	
	ECR327		18	2320(50°C)	165	76	2.2	-14	
10	ECR143H		12	2400(50°C)	462	342	1.35	-26	
	ESCOREZ	5380	85	400(148°C)	420	320	1.3	30	
	ESCOREZ	5320	125	22,000(148°C)	460	360	1.3	70	
	ESCOREZ	1310LC	93	1400(148°C)	1250	840	1.5	36	

\_

20

25

30

35

The polymers used in the present invention's hot melt adhesive generally have molecular weights between about 10,000 and 150,000, preferably between about 30,000 and about 100,000.

Further, because syndiotactic polypropylene is non-polar, the adhesives of the invention will wet olefinic substrates more effectively than the ethylene/vinyl acetate copolymer based hot melt adhesives commonly used.

The hot melt adhesives of the invention have a softening temperature of about 120°C to about 150°C, and a melting point between 90°C and 180°C, if one exists. This is a substantially higher softening temperature than the ethylene/vinyl acetate type tackifiers.

The hot melt adhesives of this invention also have good "open times" between the application of the adhesives and the formation of the joint. Good open time means sufficient time for correction of formation of a faulty joint. On solidification, the syndiotactic polypropylene based hot melt adhesive of this invention exhibits less shrinkage than corresponding isotactic polypropylene or ethylene/vinyl acetate based adhesives.

-7-

It has been found that the compositions of this invention can be heated to temperatures up to 110°C without failure of the adhesive bond. Thus, compositions are provided which have the distinct advantage of high temperature strength needed in certain applications.

5

35

While syndiotactic polypropylene homopolymer be used in the hot melt adhesive of the invention, syndiotactic polypropylene copolymers can also be used. 10 Suitable comonomers include C2and C4 to C12 alphaolefins, particularly alpha-olefin comonomers such as ethylene, 1-butene, 1-pentene, 1-hexene, 1-heptene, 1octene, particularly 1-hexene. Furthermore, the comonomers are present of from .01 to about 50 weight 15 %, based upon the weight of the copolymer, preferably 3 to 40 weight %, more preferably 6 to 20 weight %, even more preferably 12 to 17 weight %. The hot melt adhesive of the invention is particularly useful in situations where one or both of the substrates is an 20 engineering polypropylene such as two interior panels of a car. The propylene 1-hexene copolymers, also provide good bonding of dissimilar substrates. A hot melt adhesive according to this invention is typically produced by melting syndiotactic polypropylene and a 25 tackifier in a ratio of 1:1 in a stirred container. The syndiotactic polypropylene and tackifier dissolve in each other to form a viscous melt which can be applied as a melt adhesive at high temperatures. The adhesive subsequently solidifies as the temperature is decreased 30 to form a bond between adjacent substrates, such as two pieces of carpet or two automobile interior panels.

The preferred compositions of this invention include from about 40 weight percent to about 70 weight percent of a tackifying resin or combination of tackifying resins and about 30 to about 60 weight percent of syndiotactic polymer of copolymer. The more preferred compositions include about 50 to about 60

-8-

weight percent of tackifying resin and about 40 to about 50 weight percent syndiotactic polypropylene polymer. An additional and optional componemt of the composition may be an antioxidant present at about 0.1 to about 1.5 weight percent, preferrably about 0.25 to about 0.1 weight percent based upon the weight of the composition.

#### EXAMPLES 1-16

In the following examples propylene alpha-olefin polymer was blended with tackifier such that the polymer and tackifier dissolved into each other. The types, proportions and properties of blends, are identified in Table A below. The samples were then tested for SAFT, and/or PAFT and T-peel. The testing conditions and results are also in Table A below.

-9-TABLE A

																									Escorez 2520 c5/cg Resin Tg - 16°C Escorez 5380 Hydrogenated Cyclic tackify resin Tg 30°C Parapol 1300 Ethylene/butene liquid tackifiler SPP =Mw 108,000,MWD 1.6 Mp's 138°C + 127°C Tg = .06°C (3) Failure mode appears adhesive but the non adhesive side i
	S		40	2	9	3										141.0	2	THE STATE OF		285+C	0.65(3)		SF		20 c5/cg R BO Hydroge XO Ethylene/ O8,000,MW
	4		50		20	3									62 2g			28%		285+°C	0.48(3)		S.		Escorez 25: Escorez 53: Parapol 13: SPP =Mw 1
RANDOM COPOLYMER	က	20			20										27.78			H		61°C	0.05 CF		SF		
RANDOM	2	40		8											105.7°C	106.2°C		FI		118°C	0.035 CF		0.1 CF		5 wt% etyle ; isin Tg-14º
· . ·	-	50		20												117.8°C		27°C		141+°C	0.04 CF				44.7°C ropylene al sin Tg-21°C ackifying re
		CT PP•	RCP **	ECR-143H+	ECH 327++	Syndiotactic PP (SPP)	ECR 111	Escorez 2520	Escorez 5380	Parapol 1300	I PP (056)	I PP (074)	i PP (061)	SAFT'C ('C)	1" x 1" x 500g Al/Al	Opp/Opp		PAFT,C (C) AI/AI	1" x 3" x 100g (300 F/2 min)	Opp/Opp	 I-PEEL (IDS/IN) AI/AI	(300 F/2 minutes)	Офр/Орр		• CT PP of Mw 32,000 Mw/Mw 1.7 Mp = 144.7°C • RCP (PD 9282) random copolymer of propylene al 5 wt% etylene + ECR 143H Liquid Allphatic tacklifying resin Tg-21°C ++ ECR-327 Liquid Hydrogenated Cyclic tacklifying resin Tg-14°C ECR -III Aliphatic Tacklifying resin Tg 29.4°C

-9 A-

# TABLE A Continued

	16				20						50				58·C	1.7 AF	1.23 AF	
ISOTACTIC PP.	15				20					20					1120	0.46 AF	1.3 AF	O
ISOTA	4				20				20						 54°C	0.33 AF	0.7 AF	RT = Room temp. fallure AF = Adheslve fallure CF = Coheslve fallure SF = Substrate fallure X = Tested 24 hours after bonding Y = Tested at 310°F
	13			40			20	40						34°C	23.0	0.15 AF	1.3 AF	RT = Room temp. failure AF = Adhesive failure CF = Cohesive failure SF = Substrate failure X = Tested 24 hours after
	12			20				30						23C	23°C	0.5 AF	002.5 AF	RT = Roo AF = Adh CF = Coh SF = Sub X = Teste Y = Teste
	=			20			20							၁,06	84°C	2.2 AF	13.5 SF	
	5			20		20								21°C	28·C	0.18 AF	0.10 AF	9n6 3n6 ne
	6		20	20										2,98	33,C	0.28 AF	0.18 AF	city PP wt% 1-hex :6% 1-hexe 7% 1-hexe
SYNDIOTACTIC PP	æ			20	20									21°C SF	38°C SF	3.5 AF/CF	0.12 AF	CT PP = controlled tacticity PP  PP(056) = Mw 32,000;MP 123°C;6wt% 1-hexene  PP(074) = Mw=185,000;MP 114°C;6% 1-hexene  PP(061) = Mw=18,900;MP 94°C; 17% 1-hexene
SYNDIC	7	20		20										21.C	29.C	0.15 AF	0.15 AF	ontrolle Mw 32,000;  Mw=185,00 Mw=18,900
	ဖ			100														CT PP = C( IPP(056) = I IPP(074) = I IPP(061) = I

-10-

Note that when comparing Examples 4 and 9, the syndiotactic HMA (a) shows three times better PAFT than the random copolymer(4), i.e., 86°C vs 28°C. The syndiotactic HMA(9) also displayed at least 100% better PAFT than Example 3, (CT PP), as the CTPP failed at room temperature.

In addition the syndiotactic HMA of Example 7 displayed three times better T-peel than the random polypropylene blend (Ex 1), i.e. 0.15 vs. 0.04.

5

Likewise, the syndiotactic PP HMA (Ex 8) displayed three to ten times better T-peel when compared to the isotactic HMA (i.e. 3.5 vs. 0.33/0.46/1.7) (Ex.'s 14,15,16).

#### EXAMPLE 17

- Example 17 is a comparison of isotactic and syndiotactic C3/C6 copolymers containing three different levels of hexene comonomer in each case.

  These copolymers were melt blended with hydrogenated C5 tackifier (ECR-111-) in a ratio of 1:1 by weight.
- These materials are compared to a commercially available Eastman C3/C6 copolymer with 32 36% 1- hexene comonomer (See U.S. Patent 5,021,257 to Foster, issued June 4, 1971). Each of these adhesive blends were bonded after 1, 5, and 10 minutes of open time and
- 25 then debonded at 30 minutes and 24 hours after bonding.

-11-

The data are presented in Table 1.

TABLE 1

	ISOTACTIC Propylene 1-Hexene Copolymer		Eastman C3/C6 Co- polymer		otactic Lene 1-Hexene omer
% Hexene	6 17	20	32-36	6	10 17
Melt Point C	123 94	89		112	99 None
Molecular Wt.	32k 25k	19k		96k	34k 145k
Open time Set time Before Prior to Bonding Debonding	Peel streng lb/in.	th,			
1 min 30 min.	.04A 14.5C	13.6C	8.0C/A	13.5C	13.1C/A 13.4?
24 hr.	.02A 13.3C	14.4C	8.3C/A	2.0A	0 14.25C
5 min. 30 min.	0 13.4C	12.1C	9.4C/A	13.3A	13.6C/A 13.8C
24 hr.	0 2.0C	3.0C	9.3C/A	.01A	
10 min 30 min.	0 14.4C	13.3C	10.3 C/A	14.0A	
24 hr.	0 2.8A	2.3A	9.6 A	.02A	.03A 14.0C

<sup>\*</sup> All samples heated 5 minutes at 150C prior to bonding \*\* All copolymers melt blended 1:1 by weight with hydrogenated C5 tackifier

C indicates cohesive failure, A indicates adhesive failure and C/A indicates mixed mode failure. The significant findings of all of these data are:

(1) The 17 and 20% 1-hexene ISOTACTIC copolymers exhibit excellent peel strength after 10 minutes open time when debonded after 30 minutes, but the bond strength falls way off when debonded after 24 hours;

5

10

15

20

25

30

- (2) A comparable 17% 1-hexene SYNDIOTACTIC copolymer exhibits virtually identical bond strength to the isotactic initially (after 30 minutes) and maintains that bond integrity after 24 hours;
- (3) 6% hexene comonomer in the isotactic structure is insufficient to induce the slow crystallization required for long open time initially (1 minute open time is already too long) however it does appear to be sufficient in the syndiotactic structure; and
- (4) The Eastman C3/C6 blend also maintains its bond strength when debonded after 24 hours just as the syndiotactic copolymer does, however the syndiotactic copolymer blend exhibits a 55% increase in peel strength over the Eastman system.

Findings 1 and 2 are also illustrated in Figures 1 and 2 showing the 24 hour peel strength dropping in the case of the isotactic copolymer (Figure 1) and remaining around 14 lbs./in. in the case of the syndiotactic copolymer (Figure 2).

Listed in Table 2 are T-peel data comparing syndiotactic homopolymer to syndiotactic C3/C6 copolymer. These materials were melt blended with ECR-111 and bonded to aluminum, polyethylene, polypropylene, and plasticized vinyl substrates.

-13-

Samples were debonded after 2 hours.

TABLE 2

5			
		s-PP homopolymer ECR-111 (1:1 ratio by wt.)	C3/C6 copolymer ECR-111 (1:1 ratio by wt.)
			2 Hours
10	Time between bonding and debonding	2 Hours	2 hours
	T-Peel, lbs/in.		
			11.6
	Aluminum	1.95 (AF)	(CF)
15	Polyethylene	3.7 (CF)	14.25 (CF)
	Polypropylene	0.3 (AF)	20.2 (CF)

0.11

(AF)

7.5

(CF)

(AF) adhesive failure (CF) cohesive failure

Plasticized

Vinyl

25

30

35

The C3/C6 copolymer system far outperforms the homopolymer on every substrate tested. In fact, the balance of peel on this broad selection of substrates is outstanding. It is known that strong T-peel results on one substrate do not always translate to strong T-peel results on a second substrate. Applicants excellent balance lends to much greater flexibility in choosing dissimilar substates for bonding. Figure 3 and Figure 5 illustrate stress-strain curves on a C3/C6

copolymer system containing 6% hexene. Again this is a 1:1 blend with ECR-111. These were tested 2 hours and 24 hours after the pads were molded. Notice the dramatic increase in tensile stress at yield from approximately 60 psi at 2 hours to around 175 psi after 24 hours. Correspondingly, toughness increases from 167.4 to 475.6 in.-lbs./in.3 and the modulus increases from 327.5 psi to 1420 psi. These values are indicative of the very slow crystallization process inherent in these copolymers and illustrate the improved toughness of these systems with time.

10

15

20

25

30

35

Figure 4 depicts the results of a study where various controlled tacticity polypropylene (CTPP) films were heat sealed to each other with no adhesive sandwiched in the middle. Since all the previous work was done utilizing an isotactic PP film for testing, a determination of whether an isotactic PP film would, in fact, remain sealed to a syndiotactic film was made. The minimum seal temperature on the y axis indicates the lowest temperature at which a bond was formed which could not be pulled appart easily by hand. Three types of film were tested: controlled isotactic homopolymer (CiPP), controlled syndiotactic homopolymer (CsPP), and purified (deashed) syndiotactic homopolymer (psPP). Only the CiPP and the purified syndiotactic PP sealed to themselves remained sealed after 24 hours. other combinations debonded after 24 hours. The most notable finding of this study is that the purified (deashed) syndiotactic polypropylene has a 40 °F lower minimum seal temperature than its unpurified analog, illustrating the value of utilizing a purified version of the SPP polymer.

As is apparent from the foregoing description, the materials prepared and the procedures followed relate to specific embodiments of the broad invention. It is apparent from the foregoing general description and the specific embodiments that, while forms of the invention

-15-

have been illustrated and described, various modifications can be made without departing from the spirit and scope of this invention. Accordingly, it is not intended that the invention be limited thereby.

-16-

### What is claimed is:

5

20

30

35

- 1. A hot melt adhesive comprising tackifier and syndiotactic propylene homopolymer or copolymer, said homopolymer or the propylene segment of the copolymer comprising at least about 80% racemic dyads and said homopolymer or copolymer having a melting point of from 100°C to 180°C.
- 2. The hot melt adhesive of claim 1, wherein said homopolymer or propylene segment of the copolymer comprises at least 85% racemic dyads.
- 3. The hot melt adhesive of claim 1, wherein the comonomer of the copolymer is one or more of ethylene or a C<sub>4</sub> to C<sub>12</sub> alpha olefin.
  - 4. The hot melt adhesive of claim 3, wherein said copolymer comprises 1 to 50 weight % of comonomer.
  - 5. The hot melt adhesive of claim 4, wherein the comonomer is hexene-1 and is present at 3 to 20 weight %.
- 25 6. The hot melt adhesive of claim 1, wherein the tackifier is present at 1 to 70 weight %.
  - 7. The adhesive of claim 1, wherein the tackifier is a hydrogenated resin.
  - 8. A process for producing the hot melt adhesives of any of the above claims comprising combining tackifier with syndiotactic homopolymer or copolymer of propylene, said homopolymer or the propylene segment of the copolymer comprising at least about 80% racemic dyads and said polymer or copolymer further having a melting point of from 100°C to 180°C.

-17-

9. The product of any of the above claims, wherein the hot melt is characterized by a long open time.

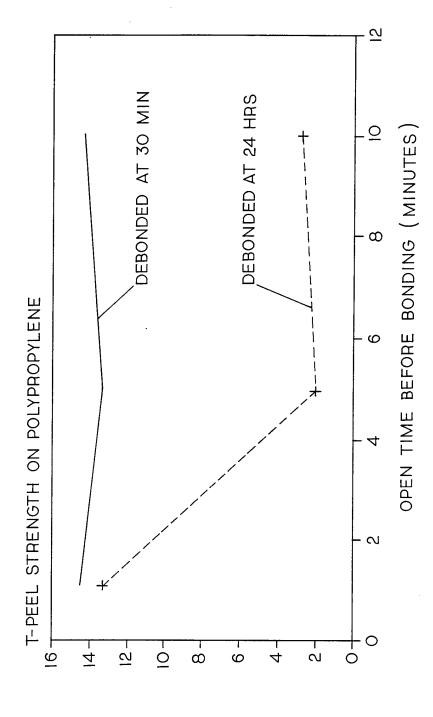
5

10. The product of any of the above claims, wherein the hot melt adhesive is characterized by a peel strength after 24 hours at or above the peel strength at 10 minutes.

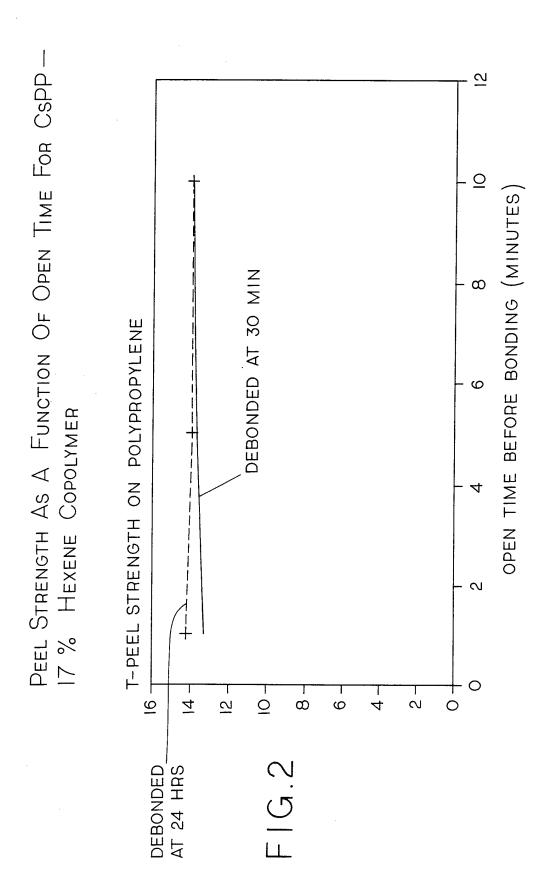
10

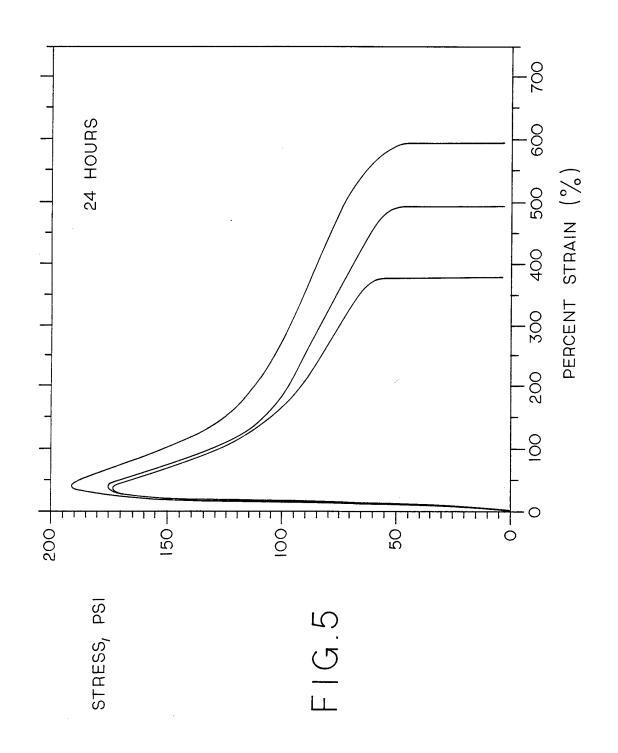
- 11. The product of any of the above claims, wherein the hot melt adhesive is characterized by an open time of about ten minutes or more.
- 12. The adhesive of any of the above claims, where the syndiotactic propylene homopolymer or copolymer is deashed.

Peel Strength As A Function Of Open Time For CIPP 17 % Hexene Copolymer

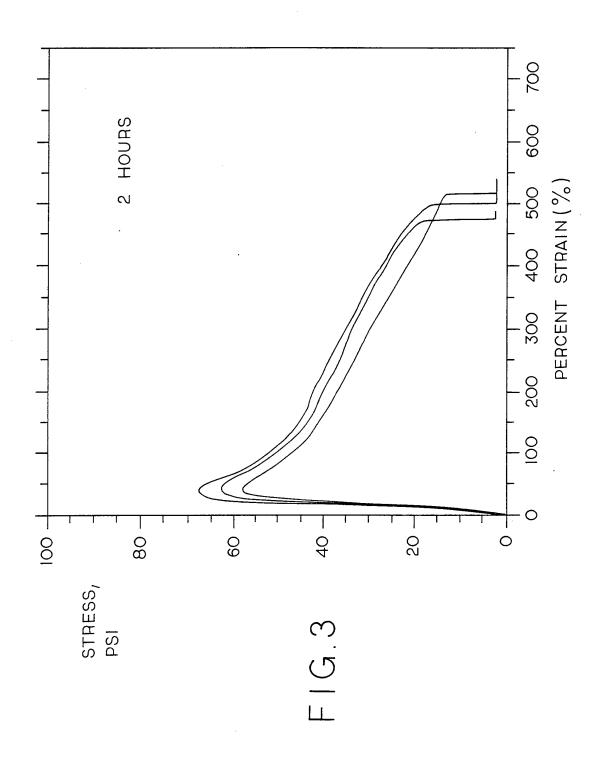


F1G.1

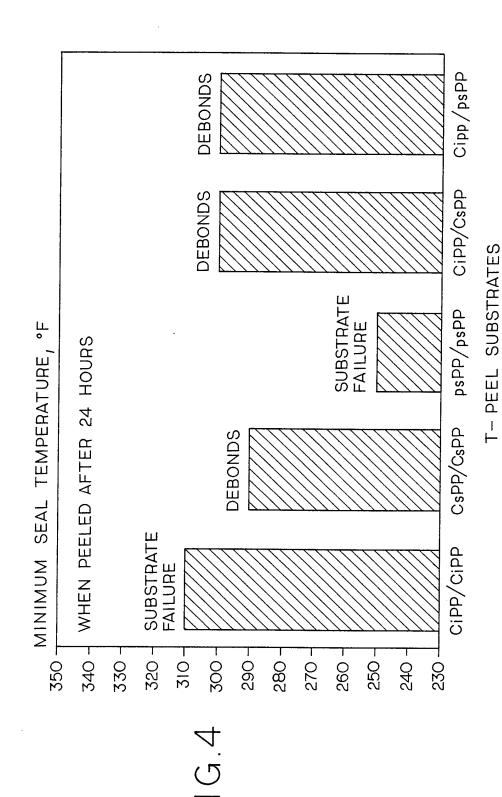




-4/5-



CTPP FILM SEALING STUDY



# INTERNATIONAL SEARCH REPORT

Inten. ..al Application No PCT/US 93/05833

A CTASS	STETCA TION OF SUBJECT AS A PROPER		
IPC 5	SIFICATION OF SUBJECT MATTER C09J123/16		
According	to International Patent Classification (IPC) or to both national class	ification and IPC	
	S SEARCHED		·
Minimum	documentation searched (classification system followed by classifica	ition symbols)	
IPC 5	CO9J	,	
Documents	ation searched other than minimum documentation to the extent that	such documents are included in the fields a	searched
Electronic o	data base consulted during the international search (name of data ba	se and, where practical, search terms used)	
C. DOCUM	MENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the r	cievant passages	Relevant to claim No.
Y	EP,A,O 350 416 (EASTMAN KODAK) 10	O January	1-12
	see page 2, line 37 - page 3, lin	ne 35	
Υ	EP,A,O 351 391 (FINA) 17 January	1990	1-12
	cited in the application	<del>)</del>	
	see page 2, line 9 - page 3, line	e 23	
A	WO,A,79 00482 (EASTMAN KODAK) 26	July 1979	1-12
	see page 3, line 1 - page 6, line	⊇ 1	
A	FR,A,1 426 429 (IMPERIAL CHEMICAL INDUSTRIES) 13 January 1965		1-12
	see page 1, column 1, last paragr page 1, column 2, paragraph 2	raph -	
i			
1	her documents are listed in the continuation of box C.	Patent family members are listed	in annex.
Special ca	tegories of cited documents:	"T" later document published after the inte	ernational filing date
	ent defining the general state of the art which is not ered to be of particular relevance	or priority date and not in conflict wi cited to understand the principle or the	th the application but seory underlying the
"E" earlier	document but published on or after the international	invention "X" document of particular relevance; the	claimed invention
"L" docume	ent which may throw doubts on priority claim(s) or	cannot be considered novel or cannot involve an inventive step when the do	t be considered to
citatio	n or other special reason (as specified)	"Y" document of particular relevance; the cannot be considered to involve an in	
O' docum other r	ent referring to an oral disclosure, use, exhibition or means	document is combined with one or m ments, such combination being obvious	ore other such docu-
"P" docume later th	ent published prior to the international filing date but han the priority date claimed	in the art.  *& document member of the same patent	· ·
Date of the	actual completion of the international search	Date of mailing of the international se	arch report
8	November 1993	<b>2</b> 4. 11. 9	3
Name and r	nailing address of the ISA  European Patent Office, P.B. 5818 Patentiaan 2	Authorized officer	
:	NL - 2280 HV Rijswijk		
	Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	SCHMIDT, H	

1

# INTERNATIONAL SEARCH REPORT

information on patent family members

Inter. nal Application No PCT/US 93/05833

				T
Patent document cited in search report	Publication date	Patent : memb		Publication date
EP-A-0350416	10-01-90	US-A- EP-A- WO-A-	4929509 0420935 9000582	29-05-90 10-04-91 25-01-90
EP-A-0351391	17-01-90	AU-B- AU-A- CN-A- JP-A-	626090 3660689 1039812 2041305	23-07-92 18-01-90 21-02-90 09-02-90
WO-A-7900482	26-07-79	US-A- CA-A- EP-A,B	4210570 1117681 0003194	01-07-80 02-02-82 25-07-79
FR-A-1426429		GB-A-	1079051	

Form PCT/ISA/210 (patent family annex) (July 1992)